

Chemistry Study Guide Stoichiometry

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Mole a unit of measurement that expresses the amount of atoms molecules or
some other unit The number of items in one mole is commonly referred to
Avogadro's number which equals 6.022×10^{23} Example One mole of carbon

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relating the changes in substances undergoing a chemical reaction This
idea is present in almost every topic in AP Chemistry so it is one of the
most important areas to study for your exam

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The mole abbreviated mol is the unit chemists use when counting numbers of
atoms or molecules in a sample The number of particles atoms molecules or
other objects in one

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separated into first term and second term general chemistry material Each
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